

AP Chemistry
Test 1 Review

Write the net ionic equation for each reaction

1. a piece of lithium metal is dropped into a container of nitrogen gas.
2. solid sodium oxide is added to water.
3. Carbon dioxide is bubbled through water
4. a solution of sodium sulfide is added to a solution of zinc nitrate.
5. ethene (ethylene) gas is bubbled through a solution of bromine.
6. solid calcium is added to warm water.
7. powdered magnesium oxide is added to a container of carbon dioxide gas.
8. hydrogen peroxide solution is heated.
9. propanol is burned completely in air.

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10. a solution of sodium hydroxide is added to a solution of ammonium chloride.

11. a strip of magnesium is added to a solution of silver nitrate.

12. solid potassium chlorate is heated.

13. dilute hydrochloric acid is added to a solution of potassium carbonate.

14. a solution of potassium iodide is added to an acidified solution of potassium dichromate.

15. Perchloric Acid is mixed with hydrogen peroxide

**THERE WILL ALSO BE QUESTIONS FROM THE TEXT BOOK ASSIGNMENT ON
THE TEST!!!!**