**Name\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_**

**For Each of the Following draw the Lewis Dot structure, name the compound then make a model of the compound, and give the name of its shape.**

1. **CS2 (Use 4 hole as center atom)**
2. **SCl4  (S breaks the octet rule) (Use 5 hole as center atom)**
3. **SCl6 (S breaks the octet rule) (Use 6 hole as center atom)**
4. **NF3 (Use 4 hole as center atom)**
5. **BF3 (B breaks the octet rule) (Use 3 hole as center atom)**
6. **SiCl4 (Use 4 hole as center atom)**
7. **H2S (Use 4 hole as center atom)**
8. **C2H4 (This is 2 of the same shapes bonded together)(Use two 4 holes as center atoms)**